Activity 2

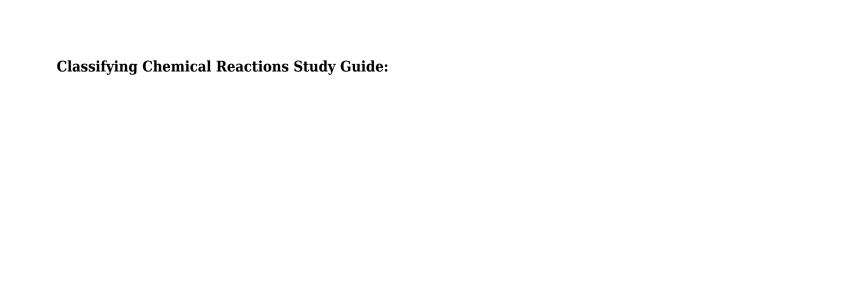
Directions. Classify the following unbalanced chemical equations according to the six types of chemical reactions. Write your answers in your notebook/ on a separate sheet of paper.

- 1. $AgNO_3 + Cu \rightarrow Cu(NO_3)_2 + Ag$
- P₄ + O₂ → P₂O₅
- HBr + Al(OH)₃ → AlBr₃ + H₂O
- 4. $O_3 \rightarrow O' + O_2$
- 5. $C_3H_6O + O_2 \rightarrow CO_2 + H_2O$

- HBr + KOH → KBr + H₂O
- H₂ + N₂ → NH₃
- 8. $Hg_2O \rightarrow Hg + O_2$
- Cl₂ + NaBr → NaCl + Br
- 10. $CH_4 + O_2 \rightarrow CO_2 + H_2O$
- 11. $SO_2 + O_2 \rightarrow SO_3$
- 12. HgO → Hg + O₂
- 13. $P + O_2 \rightarrow P_4O_{10}$
- 14. KClO₄ → KCl+ O₂
- 15. $Fe_2O_3 + C \rightarrow Fe + CO_2$
- 16. $H_2 + AgNO_3 \rightarrow Ag + HNO_3$
- CaCO₃ + HCl → CaCl₂ + H₂CO₃
- 18. $Cu + AgNO_3 \rightarrow Ag + Cu(NO_3)_2$
- 19. $Mg + O_2 \rightarrow MgO + energy$
- HCl + NaOH → NaCl + H₂O

Classifying Chemical Reactions Study Guide

PT Brinkman



The Enigmatic Realm of Classifying Chemical Reactions Study Guide: Unleashing the Language is Inner Magic

In a fast-paced digital era where connections and knowledge intertwine, the enigmatic realm of language reveals its inherent magic. Its capacity to stir emotions, ignite contemplation, and catalyze profound transformations is nothing in short supply of extraordinary. Within the captivating pages of **Classifying Chemical Reactions Study Guide** a literary masterpiece penned by a renowned author, readers attempt a transformative journey, unlocking the secrets and untapped potential embedded within each word. In this evaluation, we shall explore the book is core themes, assess its distinct writing style, and delve into its lasting affect the hearts and minds of those that partake in its reading experience.

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