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J Elliott



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web the formula for molality and molality units molality is measured in milligrammes m or milligrammes per kilogramme mol kg the molality equation is a mathematical formula that describes how much a substance weighs in this equation m moles solute per kilogramme of solution

342g of 20 by mass of Ba(OH)_2 solution sp gr 0.57 is reacted with 1200 ml of 2M HNO_3 if the final density of solution is same as pure water then molarity of the iron in resulting solution which decides the nature of the above solution is

- 1.025M
- 2.05M
- 0.888M
- none of

web what is the formula for converting molarity to molality molality is equal to the product of molarity divided by the density of the solution at that temperature $m = \frac{M}{d}$ which is more important molality or molarity both molality and molarity are concentration units and they are used interchangeably

web correct answer c both a and b q2 the density of 1 m solution of hcl is 1 0585 g ml the molality of the solution is 1 0585 1 0 10 0 0585 correct answer b 1 explanation molality begin array l m left frac 1000 m 1000d mm right end array

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web 1 molarity $m = \frac{\text{mass}}{\text{volume}}$ mass molar mass $\times 0.4500 \text{ L} = 0.825 \text{ g}$ 141.9579 g/mol $\times 0.0129 \text{ m}^3 = 0.825 \text{ g}$ 141.9579 g/mol 0.00581158 mol 0.00581158 mol 0.4500 kg 0.0129 m³ mole fraction $n_2 = \frac{0.825 \text{ g}}{141.9579 \text{ g/mol}} = 0.00581158 \text{ mol}$ $n_1 = \frac{450.0 \text{ g}}{18.015 \text{ g/mol}} = 24.97918401 \text{ mol}$

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web 1 calculate the molarity of a solution which contains 0.40 mol of a substance dissolved in 1.6 L of a solution 2 what is the molarity of a solution containing 325 g of NaCl dissolved in 750 ml of solution 3 140 g of KCl is dissolved in 600 ml of water what is the molarity 4 724 g of ammonium phosphate in 4500 ml of alcohol

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web determine the molarity for each of the following solutions 0 444 mol of CoCl_2 in 0 654 l of solution 98 0 g of phosphoric acid H_3PO_4 in 1 00 l of solution 0 2074 g of calcium hydroxide Ca(OH)_2 in 40 00 ml of solution 10 5 kg

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