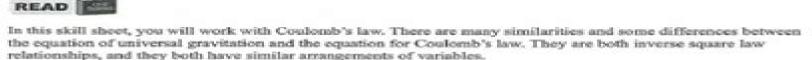
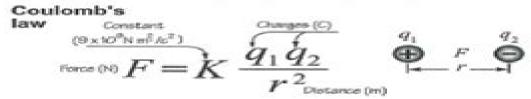
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Coulomb's Law





When two charges q_1 and q_2 are separated by a distance r_s there exists a force between them that is given by:



where F equals the force in newtons and K is a constant equal to 9×10^9 N-m²/C². The units of q_1 and q_2 are the coulombs (C). Distance is given in meters. Here are some important points about the relationships of the variables in Coulomb's law.

- Force is inversely proportional to the square of the distance between the charges. Therefore, if the distance increases by a factor of 2, the force decreases by a factor of 4.
- Force is proportional to the strength of each charge.
- When the two charges have the same sign (positive or negative), the force between them is repulsive because like charges repel.
- When the charges have opposite signs, the force between them is attractive because unlike charges attract.

PRACTICE 1

- What happens to the force between two charges if the distance between them is tripled?
- What happens to the force between two charges if the distance between them is quadrupled?
- What happens to the force between two charges if the distance between them is cut in half?
- 4. What happens to the force between two charges if the magnitude of one charge is doubled? ZF
- 5. What happens to the force between two charges is the magnitude of both charges is doubled? 44
- 6. What happens to the force between two charges if the magnitude of both charges is doubled and the distance between them is doubled? NO Charges
- 7. What happens to the force between two charges if the magnitude of both charges is doubled and the distance between them is cut in half?

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Charles W. Misner, Patrick J. Cooney

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