Name	0	Date	Class

CHAPTER 8 STUDY GUIDE

Covalent Bonding

Section 8.1 The Covalent Bond

In your textbook, read about the nature of covalent bonds.

Use each of the terms below just once to complete the passage.

covalent bond	molecule	sigma bond	exothermic	pi bond		
When sharing of el-	ectrons occurs, the attachr	nent between atom	s that results is called			
a(n) (1)	. When such					
released, and the proce	ess is (2)	, When	two or more atoms bone	by means of		
electron sharing, the re	esulting particle is called a	(n) (3)	3) If the electrons shared			
are centered between t	he two atoms, the attachm	ent is called a(n) (4)	If the		
sharing involves the or	verlap of parallel orbitals,	the attachment is c	alled a(n) (5)			
a. as single atom	t form do elements such as hydrogen, nitrogen, and oxygen normally occur? single atoms e. as molecules containing three atoms d. as molecules containing four atoms					
7. How many electry	How many electrons are shared in a double covalent bond?					
a. none	b. one	c. two	d, f	our		
a. two molecules	e distance between s of the same substance. in two attached atoms.	54 F33 F37 F17	the nuclei of two attached atoms. the orbitals of two attached atoms.			
 a. the shorter the b. the shorter the c. the shorter the 	owing relationships relationships relationships bond, the stronger the bond bond, the weaker the bond bond, the fewer the elect	nd ad rons in it				
el. the shorter the	bond, the lower the bond	dissociation energ	rivi .			

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John Krenos, Joseph Potenza, Carl Hoeger

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