

Stoichiometry Problems (continued)

3. balancing this equation
4. convert the mass of _____ of _____ substance to
5. MASS TO MO particles _____ MOLES _____ that make particles
PER ONE BALANCED COEFFICIENT
6. convert the mass _____

Problem Set One

180.0 g C_2H_6 is (1 mol C_2H_6)(30.07 g C_2H_6 /mol) (16 mol CO_2)(1 mol C_2H_6) is (480.08 g CO_2)(1 mol CO_2) is (1 L)(1.99 g) = 180.24 g CO_2

Stoichiometry of water: 180.0 g and particles ions (molar)

180.0 g C_2H_6 is (1 mol C_2H_6)(30.07 g C_2H_6) is (16 mol H_2O)(1 mol C_2H_6) is (18.02 g H_2O)(1 mol H_2O) = 18.02 g H_2O

Chemistry Note Taking Guide Episode 803

JS Bruner



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