

CP Chemistry - Zajac - Semester II Exam Study Guide / Practice Exam**Multiple Choice***Identify the choice that best completes the statement or answers the question.*

- _____ 1. All of the following are equal to Avogadro's number EXCEPT _____.
a. the number of atoms of bromine in 1 mol Br_2
b. the number of atoms of gold in 1 mol Au
c. the number of molecules of nitrogen in 1 mol N_2
d. the number of molecules of carbon monoxide in 1 mol CO
- _____ 2. How many moles of tungsten atoms are in 4.8×10^{23} atoms of tungsten?
a. 8.0×10^2 moles
b. 8.0×10^3 moles
c. 1.3×10^{-1} moles
d. 1.3×10^{-2} moles
- _____ 3. Which of the following is NOT a true about atomic mass?
a. The atomic mass is 12 g for magnesium.
b. The atomic mass is the mass of one mole of atoms.
c. The atomic mass is found by checking the periodic table.
d. The atomic mass is the number of grams of an element that is numerically equal to the mass in amu.
- _____ 4. The mass of a mole of NaCl is the _____.
a. molar mass
b. atomic mass
c. molecular mass
d. gram atomic mass
- _____ 5. What is the molar mass of AuCl_3 ?
a. 96 g
b. 150 g
c. 232.5 g
d. 303.6 g
- _____ 6. The molar mass of C_3H_8 and the molar mass of CaCO_3 contain approximately the same number of _____.
a. carbon atoms
b. cations
c. anions
d. grams
- _____ 7. How many moles of CaBr_2 are in 5.0 grams of CaBr_2 ?
a. 2.5×10^{-2} mol
b. 4.2×10^{-2} mol
c. 4.0×10^2 mol
d. 1.0×10^2 mol
- _____ 8. The lowest whole-number ratio of the elements in a compound is called the _____.
a. empirical formula
b. molecular formula
c. binary formula
d. representative formula

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