#### Lab Day 7 Purification by Acid-Base Extraction Prelab Questions

 Ust the most common organic functional groups that function as acids or bases. Write an equation of each with HCl or NaCH to show their behavior as acids or bases. (You will also enter these functional groups in the table in Question 3.)

Acids-Carboxylic Acid: R-COOH+NaOH-> B-COONa + H2O
"Phenols: Ar-OH + MOH-> Ar-ONa + H2O
"Sulfonic ocids: R-SO3H + NaOH-> B-SO3Na + H2O

Bases-Amine: R2NH+ HC1 -> R2NH2+CI Amide: R-CONH2+HC1 -> R-CONH3++CI

Fill in the following table showing the general solubility behavior of the organic acids and bases you listed above.
 Assume that they are not small enough to have significant water solubility.

Type of Organic Compound	Functional Group	Soluble in Water?	Solution in 1 M NuHCOs?	Soluble in 1 M NaOH?	Soluble in 1 M HCIP
Neutral	Many				
Acidic	Charghands	Yes	Yes	Yes	No
Weakly Acidic	Phenols	Year	Yes	Yes	Yes
Besic	Amines	No	Yes	Yes	Yes

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National Science Foundation (U.S.).

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Chemistry 121 Lab Manual Karen Eichstadt, 1995-06-01 Catalogue of Atlanta University Atlanta University, 1917 A Guide to Undergraduate Science Course and Laboratory Improvements National Science Foundation (U.S.). Directorate for Science Education, 1979 Catalogue of the Officers and Students of Atlanta University, (incorporated 1867--opened 1969) Atlanta, Ga Atlanta University, 1929 Chemical & Metallurgical Engineering Eugene Franz Roeber, Howard Coon Parmelee, 1922 The Chemical Trade Journal and Chemical Engineer G Kelville Davis, 1913 The Chemical Trade Journal and Chemical Engineer ,1913 **Pharmaceutical Record and Weekly Market Technical Manual of the American Association of Textile Chemists and Colorists** Review P. W. Bedford, 1887 The Journal of Industrial and Engineering Chemistry American Association of Textile Chemists and Colorists. 1959 ,1920

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## **Table of Contents Chem 121 Custom Lab Manual**

- 1. Understanding the eBook Chem 121 Custom Lab Manual
  - The Rise of Digital Reading Chem 121 Custom Lab Manual
  - Advantages of eBooks Over Traditional Books
- 2. Identifying Chem 121 Custom Lab Manual
  - Exploring Different Genres
  - Considering Fiction vs. Non-Fiction
  - Determining Your Reading Goals
- 3. Choosing the Right eBook Platform
  - Popular eBook Platforms
  - Features to Look for in an Chem 121 Custom Lab Manual
  - User-Friendly Interface
- 4. Exploring eBook Recommendations from Chem 121 Custom Lab Manual
  - Personalized Recommendations
  - Chem 121 Custom Lab Manual User Reviews and Ratings
  - Chem 121 Custom Lab Manual and Bestseller Lists
- 5. Accessing Chem 121 Custom Lab Manual Free and Paid eBooks
  - Chem 121 Custom Lab Manual Public Domain eBooks
  - Chem 121 Custom Lab Manual eBook Subscription Services
  - Chem 121 Custom Lab Manual Budget-Friendly Options
- 6. Navigating Chem 121 Custom Lab Manual eBook Formats

- o ePub, PDF, MOBI, and More
- Chem 121 Custom Lab Manual Compatibility with Devices
- Chem 121 Custom Lab Manual Enhanced eBook Features
- 7. Enhancing Your Reading Experience
  - o Adjustable Fonts and Text Sizes of Chem 121 Custom Lab Manual
  - Highlighting and Note-Taking Chem 121 Custom Lab Manual
  - Interactive Elements Chem 121 Custom Lab Manual
- 8. Staying Engaged with Chem 121 Custom Lab Manual
  - Joining Online Reading Communities
  - Participating in Virtual Book Clubs
  - Following Authors and Publishers Chem 121 Custom Lab Manual
- 9. Balancing eBooks and Physical Books Chem 121 Custom Lab Manual
  - Benefits of a Digital Library
  - Creating a Diverse Reading Collection Chem 121 Custom Lab Manual
- 10. Overcoming Reading Challenges
  - Dealing with Digital Eye Strain
  - Minimizing Distractions
  - Managing Screen Time
- 11. Cultivating a Reading Routine Chem 121 Custom Lab Manual
  - Setting Reading Goals Chem 121 Custom Lab Manual
  - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Chem 121 Custom Lab Manual
  - Fact-Checking eBook Content of Chem 121 Custom Lab Manual
  - Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
  - Utilizing eBooks for Skill Development
  - Exploring Educational eBooks
- 14. Embracing eBook Trends
  - Integration of Multimedia Elements
  - Interactive and Gamified eBooks

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