

CHEMISTRY EQUATIONS

Essential Tool for Chemistry Concepts, Variables and Equations Including 🦠 Sample Problems, 🔥 Common Pitfalls and Helpful Hints

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Your success in chemistry depends on your shifty to solve numerical problems.

BASIC SKILLS

CALCULATOR SURVIVAL

· Bocome familiar with your calculator before the exam; make sure you can multiply, divide, add, subtract and use all needed functions.

- Calculators never make mistakes; they take your input (intended & accidental) and give an answer.
- Look at the answer; does it make sense?

 Do a quick estimate to check your work. Sample 4.34 x 7.68 / (1.05 x 9.8) is roughly 4 x 8 / (1 x 10) or 3.2; the actual account 3.24

IOW TO DO WORD PROBLEMS

- · Read and evaluate the question before you start plugging numbers into the calculator.
- Identify the variables, constants and equations.
- Write out units of the variables and constants.
- · Work out the unit before the number.
- You may have extra information, or you may need to orbinin coontants from your text.

HOW TO WORK WITH UNITS

All numerical data has units. In chemistry we came emetrics "SIT comittee."

Pirfall: If the unit is wrong, the answer is wrong!

1. Unit prefixes: denote powers of "10" mercia A5 50°

term T. 10 C. mign G 107 killo k 103 wheel of 100° connel o 100-7 matter on 10°F. micro a 10th marke is \$10°° Formio I' borts péce p. 10°12.

Femdamental Units

- Missac kölingmen (kg):
- · Longth: motor (m)
- Temperature: Ketvin (K)
- . Timer second (td)
- * Amount of a substance: mole
- Electric charge: coulomb, (C)

Developed Units

- Area: hough suppred, m²
- Volume: length cubed, m³; 1 liter (L) = 1 dm³
- . Density: mass/splume; common unit kg/m/
- · Speed: distance/time; common unit m/s
- Electric current: ampere (A) = 1 C/s.
- Force: Newton (Nt) = 1 kg m/s²
- Emergy: Josefe (J) = 3 kg m²/s²
- Pressure: Passad (Pa) = 1 kg/km s²);

Fundamental Constants.

Will process lim person (gg/monte) Name Street o 9.109299 x 10⁻²⁴ electronic. 5 486 × 10°4 protonc 1.007276 1.67262×10^{-51}

minuteous: 1.00846-4 6.67490 × 10⁻²⁷ 1.6002 × 10⁻¹⁹ C Electronic charges

Assignation's number, NA1 6,02214 x 10²³ Ideal gas constant, R:

R, for gas calculation: 0.082 L atm/(K mol) R, for energy calculation: 8.314510 J / (K mol): Faraday Constant, 5% 96,485 Cloud.

6.626 x 10⁻¹⁴ J x Planck's Constant, let Speed of Light, et 2.9979×10" m/s

BASIC SKILLS cont.

HOW TO CONVERT DATA

The unit & numerical value are changed using a conversion factor or equation.

OW TO USE EQUATIONS OR DATA CONVERSION

Co. Sample: Convert a temperature of 45°C to T and Kelvin: Given: K = 'C + 273.15

Calculate: Temperature in K = 45°C = 273.15 - 318.15 K

Given: "F = 9/5 °C + 3.2

Calculate: Temperature in 'F = 9/5 x 45 'C + 32 -81 + 32 - 113 TF

T(K) = T(C) + 273.15 T(F) = 9/5 × T(C) + 32



OW TO USE A CONVERSION FACTOR

- Sample "I hour = 60 minutes" gives two conversion factors: 1. Multiply by "1 hour 60 minutes" to convert

 - 2. Multiply by "60 minutes/1 hour" to convert hours

Number Determine the number of boors in 45 minutes. Given: Conversion factor is "Thous 60 minutes." Calculate: Time = 45 minutes x (1 hour 60 exinotes) = 0.75 hour (minutes cancel)

Common conversions:

I angstroon (A) = 1 x 10⁻¹⁰ m: Lagt = 0.5946-4 L

Lutter = 1.003325 x; 10° Pa If person in 7950 session life.

Pitfall: Equations and conversion factors have units. Beware of inverting conversion factors.

MATH REVIEW

HOW TO WORK WITH ALGEBRAIC **EQUATIONS**

Give equal treatment to each side.

- Add or subtract
- Sample: Given, x = y, then, 4 + x = 4 + y.
- Multiply or divide:

Sample: Given, x = x, then, 4x = 4x and x/5 = x/5 Given: a-b+4, then a/(s-2)-(b+4)/(s-2), x ≠ 2.

Piefall: Dividing by some is not allowed.

HOW TO WORK WITH SCIENTIFIC NOTATION

The exponent gives the power of 10.

Sumple: 0.00045 - 4.5 x 10°5; 1345 - 1.345 x 10° Sample chemical applications: Molocule diameters are 10⁻¹⁵ m; I liter of water contains about \$x10²⁶ atoms.

MATH REVIEW cont.

HOW TO USE LOGARITHMS & EXPONENTS

- Common logarithm, logged base 7307
- Depotes number or function in powers of 10.
- Nample Gives, y = 10 ", then leggs y = 6. Natural logarithm, in: base 767(e = 2.718281829).
- Denotes number or function in powers of "g". Sample Given, $z = e^{\lambda}$, then in z = 5.
- Products: log (sy) = log s = log y
- Provers: log to "time a log too."
- Multiplication: Add exponents from each term:
 Sample: 10⁵ x 10⁵ = 10 (5-7) = 10⁸
- Division: Subtract denominator exponents from management exponents:
 - Cample 105/103- 105/30-107
- Square root: an a
- Barrierwen him or a "
- Pitfall: Your calculator has separate keys for lites (base et), logs; (base 10), 101 and et.
- Sample chemical applications: plt of acid and base.

HOW TO CALCULATE ROOTS OF A POLYNOMIAL

An equation of the form: $ax^2 + bx + c = 0$, but 2 solutions or roots, given by the quadratic formula:

potential with the second of the Ballor September 18 month of East Married 2

Sample Determine the roots for the equations $3x^2 + 4x + 1 = 0$

Given: a = 3, b = 4, c = 1

Calculate: Boot 1 = [-4 + v/4] - 4 x 3 x 1/6 / (2x3) - [-4 + ×(16-12)]/6 - (-4 + 2)/6 - -1/3 Calculate: Boot 2 = [-4 - (16-12)] 6 = (-4 - 2)/6 = -1

Pirfall: Beware round-off error. Substitute the roots. into the equation to verify results.

Sample chemical applications: Weak acids, weak bases, buffers, chemical equilibrium

HOW TO DETERMINE THE EQUATION OF A LINE

LINEAR EQUATION: y = mx + b

ms; alleger of the lines; as = AyriAre

by intercept, the line crosses the y-axis at "b"; b = y; - no; Sample Determine the equation of a line using (x,y) data.

Given: x: -2 -1 0 1 2 3 37 -2 1 4 7 10 13

Calculate: Slope = m = Aythx

-113 - (-2)(13 - (-2)) - 15/3 - 3Calculate: y-intercept = b = y₁ - mx₁

- (-2) - 3 x (-2) - -2 - 6 - 4

The equation of the line is y = 3x + 4. Sample chemical applications: Gas calculations;



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Max Fogiel, Research and Education Association

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