

# PRACTICE EXAM 2

## CHEM 1411 (Chap. 1, 2, 3, 4, 5)

- d 1. Which of the following is a physical change?
  - (a) rusting a nail
  - (b) fruits ripen
  - (c) baking a cake
  - (d) ice melts
  - (e) milk turns sour
- e 2. Which of the following is a homogeneous mixture?
  - (a) water and sand
  - (b) sand, alcohol, and sugar
  - (c) oil and vinegar
  - (d) milk of magnesium
  - (e) air
- e 3. Which of the following is a correct?
  - (a) sodium, S
  - (b) mercury, Mg
  - (c) tin, N, Sn
  - (d) potassium, P
  - (e) gold, Au
- b 4. What is the equivalent temperature for 105 Fahrenheit in Kelvin?
  - (a) 41
  - (b) 314
  - (c) 378
  - (d) 457
  - (e) -32
- a 5. A sample of carbon tetrachloride has a mass of 39.75 g and a volume of 25.0 mL. What is its density (g/mL)?
 

$d = \frac{m}{V} = \frac{39.75}{25.0 \text{ mL}}$

  - (a) 1.59
  - (b) 1.74
  - (c) 2.36
  - (d) 2.89
  - (e) 3.50
- e 6. How many significant figures are there in the measurement of  $1.0356 \times 10^4$ ?
  - (a) 1
  - (b) 2
  - (c) 3
  - (d) 4
  - (e) 5
- c 7. How many significant figures are appropriate to show in the result after carrying out the operation below?
 

$(13.7 + 0.027) + 8.221 = ?$        $\times, \div$  least s.f.       $+, -$  least decimal

$13.727$

  - (a) 1
  - (b) 2
  - (c) 3
  - (d) 4
  - (e) 5
- d 8. How many protons, electrons, and neutrons are there in  $^{130}\text{Ba}^{2+}$ ?
  - (a) 56p, 56e, 130n
  - (b) 56p, 54e, 130n
  - (c) 56p, 58e, 74n
  - (d) 56p, 54e, 74n
  - (e) 54p, 56e, 74n
- c 9. Which of the following pairs would have similar chemical and physical properties?
  - (a) Na, Mg
  - (b) H, Li
  - (c) He, Ne
  - (d) Cr, Ca
  - (e) Cl, O
- c 10. Which of the following is the molecular formula?
  - (a) NaCl
  - (b)  $\text{CsF}$
  - (c) HCl
  - (d)  $\text{AlBr}_3$
  - (e) NaCN
- e 11. Which of the following is the empirical formula?
  - (a)  $\text{C}_2\text{H}_4$
  - (b)  $\text{CH}_3\text{COOH}$
  - (c)  $\text{P}_2\text{O}_{10}$
  - (d)  $\text{NH}_4\text{NO}_3$
  - (e)  $\text{Ag}_2\text{CO}_3$
- e 12. Which of the following is a correct match?
  - (a)  $\text{BaCl}_2$ , barium dichloride
  - (b)  $\text{CS}_2$ , carbon sulfide
  - (c)  $\text{CuF}_2$ , copper (I) fluoride
  - (d)  $\text{HNO}_2$ , nitrous acid
  - (e)  $\text{MgSO}_4 \cdot 7\text{H}_2\text{O}$ , magnesium sulfate heptahydrate
- b 13. Which of the following pairs is a correct match?
  - (a)  $\text{Ca}_3(\text{PO}_4)_2$ , calcium (II) phosphate
  - (c)  $\text{Mg}_2\text{O}_3$ , magnesium oxide
  - (b)  $(\text{NH}_4)_2\text{SO}_4$ , ammonium sulfate
  - (e)  $\text{H}_2\text{CO}_3$ , carbon acid
  - (d)  $\text{Cu}(\text{NO}_3)_2$ , copper dinitrate

# Chem 1411 Lab Manual Answer Key

**DJ Losen**



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